



**SANTOSH**  
*Academia*  
IIT-JEE | NEET | Foundation

## Answers & Solutions

Time : 3 hrs.

*for*

M.M. : 300

# JEE (Main)-2025 Phase-1

## [Computer Based Test (CBT) mode]

### (Mathematics, Physics and Chemistry)

28/01/2025

Morning

#### IMPORTANT INSTRUCTIONS:

- (1) The test is of **3 hours** duration.
- (2) This test paper consists of 75 questions. Each subject (MPC) has 25 questions. The maximum marks are 300.
- (3) This question paper contains **Three** Parts. **Part-A** is Physics, **Part-B** is Chemistry and **Part-C** is **Mathematics**. Each part has only two sections: **Section-A** and **Section-B**.
- (4) **Section - A** : Attempt all questions.
- (5) **Section - B** : Attempt all questions.
- (6) **Section - A (01 – 20)** contains 20 multiple choice questions which have **only one correct answer**. Each question carries **+4 marks** for correct answer and **-1 mark** for wrong answer.
- (7) **Section - B (21 – 25)** contains 5 **Numerical value** based questions. The answer to each question should be rounded off to the **nearest integer**. Each question carries **+4 marks** for correct answer and **-1 mark** for wrong answer.



**SANTOSH**  
Deemed to be University

Head Office: Santosh University, No 1, Santosh Nagar, Pratap Vihar, Ghaziabad

Ghaziabad Branch : J-10, Aniket Arcade, RDC Ghaziabad ☎ +91 93114 41940, +91 93114 42359



**MATHEMATICS**

**SECTION - A**

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

1. Let  $\langle a_n \rangle$  be a sequence such that  $a_0 = 0$ ,  $a_1 = \frac{1}{2}$  and  $2a_{n+2} = 5a_{n+1} - 3a_n$ ,  $n = 0, 1, 2, 3, \dots$ . Then  $\sum_{k=1}^{100} a_k$

is equal to

- (1)  $3a_{100} - 100$                       (2)  $3a_{100} + 100$   
(3)  $3a_{99} - 100$                       (4)  $3a_{99} + 100$

**Answer (1)**

**Sol.**  $2a_{n+2} = 5a_{n+1} - 3a_n$ ,  $n = 0, 1, 2, \dots$  using characteristic equation

$$2x^2 = 5x - 3$$

$$\Rightarrow x = 1, \frac{3}{2}$$

$$\Rightarrow a_n = p(1)^n + q\left(\frac{3}{2}\right)^n$$

$$\Rightarrow a_0 = p + q = 0$$

$$a_1 = p + \frac{3}{2}q = \frac{1}{2} \Rightarrow q = 1, p = -1$$

$$\Rightarrow a_n = \left(\frac{3}{2}\right)^n - 1$$

$$\sum_{k=1}^{100} a_k = \sum_{k=1}^{100} \left(\frac{3}{2}\right)^k - \sum_{k=1}^{100} (1)$$

$$= \frac{\left(\frac{3}{2}\right)\left[\left(\frac{3}{2}\right)^{100} - 1\right]}{\left(\frac{3}{2} - 1\right)} - 100$$

$$= 3\left[\left(\frac{3}{2}\right)^{100} - 1\right] - 100$$

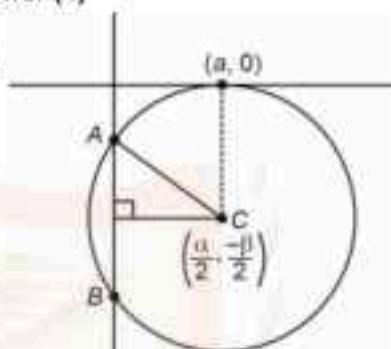
$$= 3a_{100} - 100$$

2. Let the equation of the circle, which touches  $x$ -axis at the point  $(a, 0)$ ,  $a > 0$  and cuts off an intercept of length  $b$  on  $y$ -axis be  $x^2 + y^2 - \alpha x + \beta y + \gamma = 0$ . If the circle lies below  $x$ -axis, then the ordered pair  $(2a, b^2)$  is equal to

- (1)  $(\alpha, \beta^2 + 4\gamma)$                       (2)  $(\gamma, \beta^2 - 4\alpha)$   
(3)  $(\gamma, \beta^2 + 4\alpha)$                       (4)  $(\alpha, \beta^2 - 4\gamma)$

**Answer (4)**

**Sol.**



$$r = \frac{AB}{2} = \frac{b}{2}$$

$$a = \frac{\alpha}{2} \Rightarrow 2a = \alpha$$

$$\Rightarrow r^2 = \left(\frac{b}{2}\right)^2 + \left(\frac{a}{2}\right)^2$$

$$\Rightarrow \frac{\beta^2}{4} = \frac{b^2}{4} + a^2$$

$$\Rightarrow b^2 = \beta^2 - 4a^2 = \beta^2 - (2a)^2 = \beta^2 - \alpha^2$$

$$r^2 = \frac{\beta^2}{4} = \frac{\alpha^2}{4} + \frac{\beta^2}{4} - \gamma \Rightarrow \gamma = \frac{\alpha^2}{4}$$

$$\Rightarrow b^2 = \beta^2 - 4\gamma$$

$$(2a, b^2) = (\alpha, \beta^2 - 4\gamma)$$

3. Two number  $k_1$  and  $k_2$  are randomly chosen from the set of natural numbers. Then, the probability that the value of  $k_1 + k_2$ , ( $i = \sqrt{-1}$ ) is non-zero, equals

- (1)  $\frac{2}{3}$     (2)  $\frac{1}{4}$   
(3)  $\frac{3}{4}$     (4)  $\frac{1}{2}$

**Answer (3)**



Sol.  $20 \sum_{r=1}^{25} T_r = 20 \left[ \frac{25}{2} \left[ a + \frac{1}{20} \right] \right] = 13$

$$\Rightarrow a = \frac{1}{20 \times 25}$$

$$\therefore T_{25} = a + 24d = \frac{1}{20}$$

$$\Rightarrow d = \frac{1}{20 \times 25}$$

$$T_m = a + (m-1)d = \frac{1}{25}$$

$$= \frac{1}{20 \times 25} m = \frac{1}{25} \Rightarrow m = 20$$

Now,  $5m \sum_{r=m}^{2m} T_r = 5 \times 20 \left[ \sum_{r=20}^{40} T_r \right]$

$$= 100 \left[ \frac{40}{2} (2a + 39d) - \frac{19}{2} (2a + 18d) \right]$$

$$\therefore a = d$$

$$\Rightarrow 100 \left[ \frac{40}{2} \times 41d - \frac{19}{2} \times 21d \right]$$

$$= 126$$

7. Let O be the origin, the point A be  $z_1 = \sqrt{3} + 2\sqrt{2}i$ , the point B( $z_2$ ) be such that  $\sqrt{3}|z_2| = |z_1|$  and

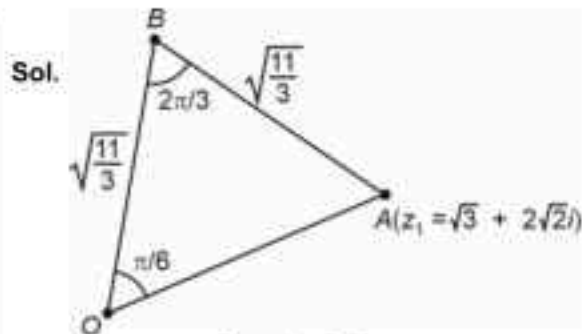
$$\arg(z_2) = \arg(z_1) + \frac{\pi}{6}. \text{ Then}$$

- (1) ABO is a scalene triangle  
(2) ABO is an obtuse angled isosceles triangle

(3) area of triangle ABO is  $\frac{11}{\sqrt{3}}$

(4) area of triangle ABO is  $\frac{11}{4}$

Answer (2)



$$OA = |z_1| = \sqrt{3+8} = \sqrt{11}$$

$$\text{and } OB = \frac{1}{\sqrt{3}} |z_1| = \frac{\sqrt{11}}{\sqrt{3}}$$

$$AB^2 = OA^2 + OB^2 - 2 \cdot OA \cdot OB \cos \frac{\pi}{6}$$

$$= 11 + \frac{11}{3} - 2 \cdot \frac{11}{\sqrt{3}} \cdot \frac{\sqrt{3}}{2}$$

$$\therefore AB = \frac{\sqrt{11}}{\sqrt{3}}$$

$$\therefore \text{Area of } \triangle ABD = \frac{1}{2} OA \cdot OB \cdot \sin \frac{\pi}{6}$$

$$= \frac{11}{4\sqrt{3}} \text{ sq. units}$$

$$\text{Here } OB = AB \text{ and } \angle A = \frac{2\pi}{3}$$

$\therefore \triangle ABD$  is an obtuse angled isosceles triangle.

8. Three defective oranges are accidentally mixed with seven good ones and on looking at them, it is not possible to differentiate between them. Two oranges are drawn at random from the lot. If  $x$  denotes the number of defective oranges, then the variance of  $x$  is
- (1) 18/25                      (2) 26/75  
(3) 28/75                      (4) 14/25

Answer (3)

Sol. There are 3 bad oranges and 7 good oranges.

$\therefore X =$  number of bad oranges drawn.

X	0	1	2
P(X)	$\frac{{}^7C_2}{{}^{10}C_2}$	$\frac{{}^3C_1 \times {}^7C_1}{{}^{10}C_2}$	$\frac{{}^3C_2}{{}^{10}C_2}$

∴ Variance

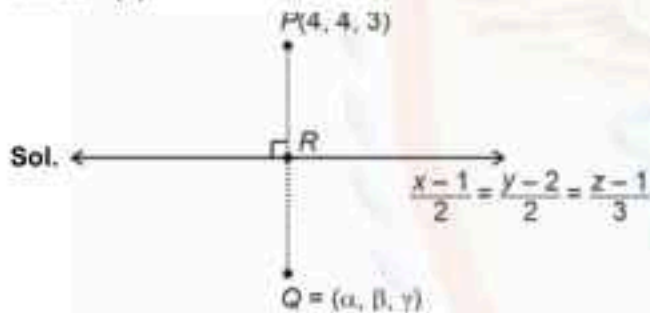
$$= 0^2 \cdot \frac{{}^7C_2}{{}^{10}C_2} + 1^2 \cdot \left( \frac{3 \times 7}{{}^{10}C_2} \right) + 2^2 \left( \frac{3}{{}^{10}C_2} \right) - \left( 0 + 1 \cdot \frac{3 \times 7}{{}^{10}C_2} + 2 \cdot \frac{3}{{}^{10}C_2} \right)^2$$

$$= \frac{28}{75}$$

9. If the image of the point (4, 4, 3) in the line  $\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{3}$  is  $(\alpha, \beta, \gamma)$ , then  $\alpha + \beta + \gamma$  is equal to

- (1) 12 (2) 7  
(3) 9 (4) 8

Answer (3)



Sol. Let coordinate of  $R = (2r + 1, r + 2, 3r + 1)$   
∴  $PR$  is perpendicular to given line.  
∴  $(2r - 3) \cdot 2 + (r - 2) \cdot 1 + (3r - 2) \cdot 3 = 0$   
∴  $r = 1$   
∴ Coordinate of  $R = (3, 3, 4)$   
∴  $(\alpha, \beta, \gamma) = (2, 2, 5)$   
∴  $\alpha + \beta + \gamma = 9$

10.  $\cos \left( \sin^{-1} \frac{3}{5} + \sin^{-1} \frac{5}{13} + \sin^{-1} \frac{33}{65} \right)$  is equal to:

- (1) 1 (2)  $\frac{33}{65}$   
(3)  $\frac{32}{65}$  (4) 0

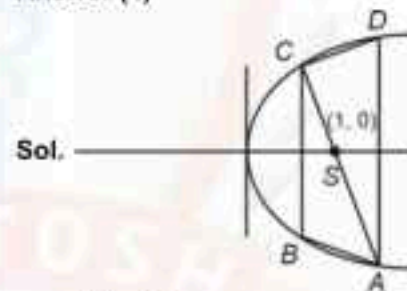
Answer (4)

Sol.  $\cos \left( \sin^{-1} \frac{3}{5} + \sin^{-1} \left[ \frac{5}{13} \sqrt{1 - \frac{33^2}{65^2}} + \frac{33}{65} \sqrt{1 - \frac{5^2}{13^2}} \right] \right)$   
=  $\cos \left( \sin^{-1} \frac{3}{5} + \sin^{-1} \frac{4}{5} \right)$   
=  $\cos \left( \sin^{-1} \frac{3}{5} + \cos^{-1} \frac{3}{5} \right)$   
=  $\cos \left( \frac{\pi}{2} \right) = 0$

11. Let  $ABCD$  be a trapezium whose vertices lie on the parabola  $y^2 = 4x$ . Let the sides  $AD$  and  $BC$  of the trapezium be parallel to  $y$ -axis. If the diagonal  $AC$  is of length  $\frac{25}{4}$  and it passes through the point (1, 0), then the area of  $ABCD$  is

- (1)  $\frac{125}{8}$  (2)  $\frac{25}{2}$   
(3)  $\frac{75}{8}$  (4)  $\frac{75}{4}$

Answer (4)



Sol.

Take  $AS = c$   
 $\frac{1}{AS} + \frac{1}{CS} = \frac{1}{a}$   
 $\Rightarrow \frac{1}{c} + \frac{1}{\frac{25}{4} - c} = \frac{1}{a}$   
 $\Rightarrow 4c^2 - 25c + 25 = 0$   
 $\Rightarrow c = \frac{5}{4}, 5$   
 $AS = c = 1 + t^2$   
 $c = \frac{5}{4} \Rightarrow t = \pm \frac{1}{2}$   
 $c = 5 \Rightarrow t = \pm 2$

$$(at^2, 2at) = A\left(\frac{1}{4}, 1\right), D\left(\frac{1}{4}, -1\right), B(4, 4)$$

$$C(4, -4) \Rightarrow AD = 2, BC = 8$$

$$\begin{aligned} \text{Area of trapezium} &= \frac{1}{2}(2+8) \times \frac{15}{4} \\ &= \frac{75}{4} \end{aligned}$$

12. If  $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \frac{96x^2 \cos^2 x}{1+e^x} dx = \pi(\alpha\pi^2 + \beta), \alpha, \beta \in \mathbb{Z}$ ,

then  $(\alpha + \beta)^2$  equals

- (1) 100                      (2) 196  
(3) 144                      (4) 64

**Answer (1)**

**Sol.**  $I = \int_0^{\frac{\pi}{2}} \frac{96x^2 \cos^2 x}{1+e^x} dx$

$$2I = 2 \int_0^{\frac{\pi}{2}} 96x^2 \cos^2 x dx$$

$$I = 96 \int_0^{\frac{\pi}{2}} x^2 \cos^2 x dx$$

$$= 48 \int_0^{\frac{\pi}{2}} x^2 (1 + \cos 2x) dx$$

$$= 2\pi^2 + 48(0 - 0) - 48 \int_0^{\frac{\pi}{2}} x \sin 2x dx$$

$$= 2\pi^2 - 12\pi + [0 - 0] = \pi(2\pi^2 - 12)$$

$$= \pi(\alpha\pi^2 + \beta)$$

$$\Rightarrow (\alpha + \beta)^2 = 100$$

13. The sum of all local minimum values of the function

$$f(x) = \begin{cases} 1-2x, & x < -1 \\ \frac{1}{3}(7+2|x|), & -1 \leq x \leq 2 \\ \frac{11}{18}(x-4)(x-5), & x > 2 \end{cases} \quad \text{is}$$

(1)  $\frac{131}{72}$

(2)  $\frac{167}{72}$

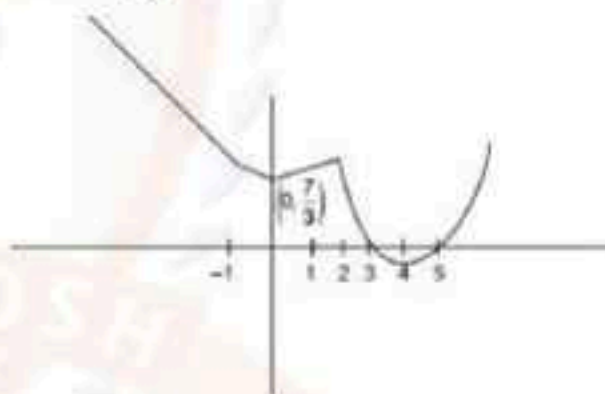
(3)  $\frac{171}{72}$

(4)  $\frac{157}{72}$

**Answer (4)**

**Sol.**  $f(x) = \begin{cases} 1-2x, & x < -1 \\ \frac{1}{3}(7+2|x|), & -1 \leq x \leq 2 \\ \frac{11}{18}(x-4)(x-5), & x > 2 \end{cases}$

$$\Rightarrow f(x) = \begin{cases} 1-2x, & x < -1 \\ \frac{1}{3}(7-2x), & -1 \leq x < 0 \\ \frac{1}{3}(7+2x), & 0 \leq x \leq 2 \\ \frac{11}{8}(x-4)(x-5), & x > 2 \end{cases}$$



Minimum value at  $x = 0$  and  $x = 4.5$

at  $x = 0, f(0) = \frac{7}{3}$

at  $x = 4.5, f(4.5) = \frac{-11}{72}$

$\Rightarrow$  Total sum =  $\frac{7}{3} - \frac{11}{72}$

=  $\frac{157}{72}$

$\Rightarrow$  Option (4) is correct

14. Let  $A(x, y, z)$  be a point in  $xy$ -plane, which is equidistant from the points  $(0, 3, 2)$ ,  $(2, 0, 3)$  and  $(0, 0, 1)$ .

Let  $B = (1, 4, -1)$  and  $C = (2, 0, -2)$ . Then among the statements

(S1) :  $\triangle ABC$  is an isosceles right angled triangle, and

(S2) : The area of  $\triangle ABC$  is  $\frac{9\sqrt{2}}{2}$ .

- (1) Both are true                      (2) Only (S1) is true  
 (3) Only (S2) is true                (4) Both are false

**Answer (2)**

**Sol.**  $A(x, y, z)$      $P(0, 3, 2)$ ,  $Q(2, 0, 3)$ ,  $R(0, 0, 1)$

∴

As  $A$  is in  $xy$ -plane

$$\Rightarrow A(x, y, 0)$$

$$AP^2 = AQ^2$$

$$(x-0)^2 + (y-3)^2 + (0-2)^2 = (x-2)^2 + (y-0)^2 + (0-3)^2$$

$$x^2 + y^2 - 6y + 9 + 4 = x^2 - 4x + 4 + y^2 + 9$$

$$\Rightarrow -6y = -4x$$

$$\Rightarrow 2x = 3y \quad \dots (i)$$

$$AP^2 = AR^2$$

$$x^2 + y^2 - 6y + 9 + 4 = x^2 + y^2 + 1$$

$$\Rightarrow y = 2$$

$$\Rightarrow x = 3$$

$$A(3, 2, 0), B(1, 4, -1), C(2, 0, -2)$$

$$\vec{AB} = -2\hat{i} + 2\hat{j} - \hat{k}$$

$$\vec{BC} = \hat{i} - 4\hat{j} - \hat{k}$$

$$\vec{AC} = -\hat{i} - 2\hat{j} - 2\hat{k}$$

$$\vec{AB} \cdot \vec{AC} = 2 - 4 + 2 = 0$$

$\Rightarrow \triangle ABC$  is right angle triangle at  $A$

$$\text{Also, } AB = 3, BC = 3\sqrt{2}, AC = 3$$

$$\Rightarrow AB = AC$$

$$\text{ar}(\triangle ABC) = \frac{1}{2} \times AB \times AC = \frac{9}{2}$$

$\Rightarrow$  only (S1) is true

$\Rightarrow$  option (2) is correct

15. The relation  $R = \{(x, y) : x, y \in \mathbb{Z} \text{ and } x + y \text{ is even}\}$  is:

- (1) Reflexive and transitive but not symmetric  
 (2) Symmetric and transitive but not reflexive  
 (3) Reflexive and symmetric but not transitive  
 (4) An equivalence relation

**Answer (4)**

**Sol.** For reflexive  $(x, x) \in R, x \in \mathbb{Z}$

$$\Rightarrow x + x = 2x \rightarrow \text{even}$$

For symmetric of  $(x, y) \in R$  then  $(y, x) \in R$  when  $x, y \in \mathbb{Z}$

$$x + y \rightarrow \text{even}$$

$$\Rightarrow y + x \rightarrow \text{even}$$

for transitive if  $(x, y) \in R \Rightarrow x + y \rightarrow \text{even}$

$$(y, z) \in R \Rightarrow y + z \rightarrow \text{even}$$

$$x + 2y + z \rightarrow \text{even}$$

$$\Rightarrow x + z \text{ is even}$$

$$\Rightarrow (x, z) \in R$$

$\Rightarrow R$  is an equivalence relation.

$\Rightarrow$  Option (4) is correct.

16. Let  $f: \mathbb{R} \rightarrow \mathbb{R}$  be a function defined by

$$f(x) = (2 + 3a)x^2 + \left(\frac{a+2}{a-1}\right)x + b, a \neq 1. \text{ If } f(x+y) =$$

$$f(x) + f(y) + 1 - \frac{2}{7}xy, \text{ then the value of } 28 \sum_{i=1}^a |f(i)| \text{ is}$$

- (1) 715                                      (2) 675  
 (3) 735                                      (4) 545

**Answer (2)**

**Sol.** Put  $y = 0$

$$f(x) = f(0) + f(x) + 1 - 0$$

$$f(0) = -1$$

$$f(0) = 0 + 0 + b$$

$$\Rightarrow b = -1$$

$$f(-1 + 1) = f(-1) + f(1) + 1 + \frac{2}{7}$$

$$f(0) = f(-1) + f(1) + \frac{9}{7}$$

$$-1 = (2+3a) + \left(\frac{a+2}{a-1}\right)(-1) + b + (2+3a)$$

$$+ \frac{a+2}{a-1} + b + \frac{9}{7}$$

$$-1 = 4 + 6a - 2 + \frac{9}{7}$$

$$-1 = 2 + \frac{9}{7} + 6a$$

$$6a = -1 - 2 - \frac{9}{7}$$

$$a = \frac{-5}{7}$$

$$f(x) = \frac{-x^2}{7} + \frac{9}{7}x - 1$$

$$f(x) = \frac{-x^2}{7} - \frac{3}{4}x - 1$$

$$\sum_{i=1}^5 f(i) = -\frac{1}{7} \left( \frac{5 \times 6 \times 11}{6} \right) - \frac{3}{4} \left( \frac{5 \times 6}{2} \right) - 5$$

$$= \frac{-55}{7} - \frac{45}{4} - 5$$

$$= \frac{675}{28}$$

$$\Rightarrow 28 \left| \sum_{i=1}^5 f(i) \right| = 675$$

17. The area (in sq. units) of the region

$$\{(x, y) : 0 \leq y \leq 2|x| + 1, 0 \leq y \leq x^2 + 1, |x| \leq 3\}$$

(1)  $\frac{17}{3}$

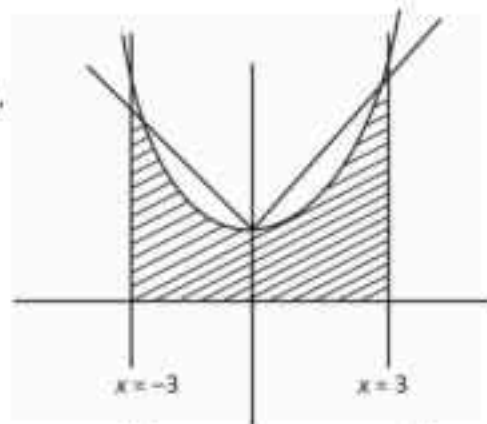
(2)  $\frac{80}{3}$

(3)  $\frac{64}{3}$

(4)  $\frac{32}{3}$

**Answer (3)**

**Sol.**



$$\text{Area} = 2 \left[ \int_0^3 (x^2 + 1) dx + \frac{1}{2} [5 + 7] \times 3 \right]$$

$$= \frac{64}{3}$$

18. If  $f(x) = \frac{2^x}{2^x + \sqrt{2}}$ ,  $x \in \mathbb{R}$ , then

$\sum_{k=1}^{81} f\left(\frac{k}{82}\right)$  is equal to

(1)  $81\sqrt{2}$  (2) 82

(3)  $\frac{81}{2}$  (4) 41

**Answer (3)**

**Sol.**  $f(x) = \frac{2^x}{2^x + 2^{1/2}} = \frac{2^x}{2^x + \sqrt{2}}$

$$f(1-x) = \frac{2^{1-x}}{2^{1-x} + 2^{1/2}} = \frac{\frac{2}{2^x}}{\frac{2}{2^x} + 2^{1/2}} = \frac{2}{2 + \sqrt{2} 2^x}$$

$$= \frac{\sqrt{2}}{2^x + \sqrt{2}}$$

$$\Rightarrow f(x) + f(1-x) = \frac{\sqrt{2} + 2^x}{\sqrt{2} + 2^x} = 1$$

$$\Rightarrow \sum_{k=1}^{81} f\left(\frac{k}{82}\right) + f\left(\frac{2}{82}\right) + f\left(\frac{3}{82}\right) + \dots$$

$$+ \dots + f\left(\frac{40}{82}\right) + f\left(\frac{41}{82}\right) + f\left(\frac{42}{82}\right)$$

$$+ \dots + f\left(\frac{79}{82}\right) + f\left(\frac{80}{82}\right) + f\left(\frac{81}{82}\right)$$





22. Let  $M$  denote the set of all real matrices of order  $3 \times 3$  and let  $S = \{-3, -2, -1, 1, 2\}$ . Let

$$S_1 = \{A = [a_{ij}] \in M : A = A^T \text{ and } a_{ij} \in S, \forall i, j\},$$

$$S_2 = \{A = [a_{ij}] \in M : A = -A^T \text{ and } a_{ij} \in S, \forall i, j\},$$

$$S_3 = \{A = [a_{ij}] \in M : a_{11} + a_{22} + a_{33} = 0 \text{ and } a_{ij} \in S, \forall i, j\}.$$

If  $n(S_1 \cup S_2 \cup S_3) = 125\alpha$ , then  $\alpha$  equals \_\_\_\_\_.

**Answer (1613)**

Sol. 
$$\begin{bmatrix} a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33} \end{bmatrix}$$

Number of elements in  $S_1 : A = A^T \Rightarrow 5^3 \times 5^3$

Number of elements in  $A = -A^T \Rightarrow 0$

Similarly, number of elements in  $S_3 \Rightarrow$

$$\left. \begin{array}{l} a_{11} + a_{22} + a_{33} = 0 \Rightarrow (1, 2, -3) \Rightarrow 3! \\ \text{or } (1, 1, -2) \Rightarrow 3 \\ \text{or } (-1, -1, 2) \Rightarrow 3 \end{array} \right\} = 12 \times 5^9$$

$$n(S_1 \cap S_2) = 12 \times 5^9$$

$$n(S_1 \cup S_2 \cup S_3) = 5^6(1 + 12) - 12 \times 5^9$$

$$\Rightarrow 5^9 \times [13 \times 5^3 - 12] = 125\alpha$$

$$\alpha = 1613$$

23. Let  $\vec{a} = \hat{i} + \hat{j} + \hat{k}, \vec{b} = 2\hat{j} + 2\hat{j} + \hat{k}$  and  $\vec{d} = \vec{a} \times \vec{b}$ . If  $\vec{c}$  is a vector such that  $\vec{a} \cdot \vec{c} = |\vec{c}|, |\vec{c} - 2\vec{a}|^2 = 8$  and the angle between  $\vec{d}$  and  $\vec{c}$  is  $\frac{\pi}{4}$ , then

$$|10 - 3\vec{b} \cdot \vec{c}| + |\vec{d} \times \vec{c}|^2 \text{ is equal to } \underline{\hspace{2cm}}$$

**Answer (6)**

Sol.  $\vec{a} = \hat{i} + \hat{j} + \hat{k}$

$$\vec{b} = 2\hat{j} + 2\hat{j} + \hat{k}$$

$$\vec{d} = \vec{a} \times \vec{b}$$

$$= -\hat{i} + \hat{j}$$

$$|\vec{c} - 2\vec{a}|^2 = 8$$

$$|\vec{c}|^2 + 4|\vec{a}|^2 - 4\vec{a} \cdot \vec{c} = 8$$

$$|\vec{c}|^2 + 12 - 4|\vec{c}| = 8$$

$$|\vec{c}|^2 - 4|\vec{c}| + 4 = 0$$

$$|\vec{c}|^2 = 2$$

$$\vec{d} = \vec{a} \times \vec{b}$$

$$\vec{d} \times \vec{c} = (\vec{a} \times \vec{b}) \times \vec{c}$$

$$\left( |\vec{d}| \times |\vec{c}| \sin \frac{\pi}{4} \right)^2 = \left( (\vec{a} \cdot \vec{c})\vec{b} - (\vec{b} \cdot \vec{c})\vec{a} \right)^2$$

$$4 = 4|\vec{b}|^2 + (\vec{b} \cdot \vec{c})2(|\vec{a}|^2) - 2(\vec{b} \cdot \vec{c})(\vec{a} \cdot \vec{b})$$

$$\text{Let } \vec{b} \cdot \vec{c} = x$$

$$4 = 36 + 3x^2 - 20x$$

$$3x^2 - 20x + 32 = 0$$

$$x = \frac{8}{3}, 4$$

$$\Rightarrow \vec{b} \cdot \vec{c} = \frac{8}{3}, 4$$

$$\Rightarrow \vec{b} \cdot \vec{c} = \frac{8}{3}$$

$$\text{Now, } |10 - 3\vec{b} \cdot \vec{c}| + |\vec{d} \times \vec{c}|^2$$

$$= |10 - 8| + (2)^2 = 6$$

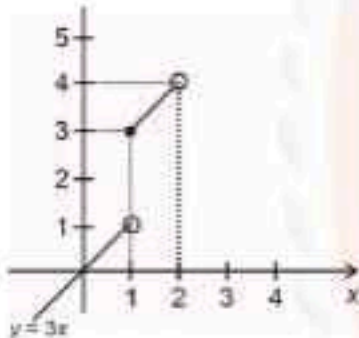


24. Let  $f(x) = \begin{cases} 3x & ; x < 0 \\ \min\{1+x[x], x+2[x]\} & ; 0 \leq x \leq 2 \\ 5 & ; x > 2 \end{cases}$

where  $[ \cdot ]$  denotes greatest integer function. If  $\alpha$  and  $\beta$  are the number of points, where  $f$  is not continuous and is not differentiable, respectively, then  $\alpha + \beta$  equals \_\_\_\_\_.

**Answer (5)**

**Sol.**  $f(x) = \begin{cases} 3x & ; x < 0 \\ \min\{1+x, x\} & ; 0 \leq x < 1 \\ \min\{2+x, x+2\} & ; 1 \leq x < 2 \\ 5 & ; x > 2 \end{cases}$



$$f(x) = \begin{cases} 3x & ; x < 0 \\ x & ; 0 \leq x < 1 \\ x+2 & ; 1 \leq x < 2 \\ 5 & ; x > 2 \end{cases}$$

Not continuous at  $x \in \{0, 2\} \Rightarrow \alpha = 2$

Not differentiable at  $x \in \{0, 1, 2\} \Rightarrow \beta = 3$

$\therefore \alpha + \beta = 5$

25. Let  $E_1 : \frac{x^2}{9} + \frac{y^2}{4} = 1$  be an ellipse. Ellipses  $E_i$ 's is constructed such that their centres and eccentricities are same as that of  $E_1$ , and length of minor axis of  $E_i$  is the length of major axis of  $E_{i-1}$  ( $i \geq 1$ ). If  $A_i$  is the area of the ellipse  $E_i$ , then  $\frac{5}{\pi} \left( \sum_{i=1}^{\infty} A_i \right)$ , is equal to \_\_\_\_\_.

**Answer (54)**

**Sol.**  $E_1 : \frac{x^2}{9} + \frac{y^2}{4} = 1$

Let  $b_i$  be minor axis of  $E_i$  and  $a_i$  be major axis of  $E_i$ .

$$\Rightarrow e_i = 1 - \frac{b_i^2}{a_i^2}$$

Now,  $b_{i+1}$  be minor axis of  $E_{i+1}$  and  $a_{i+1}$  be major axis of  $E_{i+1}$ .

$$\Rightarrow e_{i+1} = 1 - \frac{b_{i+1}^2}{a_{i+1}^2}$$

Also  $a_{i+1} = b_i$  and  $e_i = e_{i+1}$

$$\Rightarrow \frac{b_i^2}{a_i^2} = \frac{b_{i+1}^2}{b_i^2}$$

$$\Rightarrow b_{i+1} = \frac{b_i^2}{a_i}$$

$\Rightarrow$  Area of  $E_i = S_i = \pi a_i b_i$

$$\Rightarrow S_{i+1} = \pi a_{i+1} b_{i+1}$$

$$= \pi (b_i) \left( \frac{b_i^2}{a_i} \right)$$

$$= \pi (b_i a_i) \left( \frac{b_i}{a_i} \right)^2$$

$$S_{i+1} = S_i (1 - e_i^2)$$

$$S_{i+1} = S_i \left( 1 - \left( 1 - \frac{4}{9} \right) \right) = S_i \left( \frac{4}{9} \right)$$

$$\Rightarrow S_1 = 6\pi, S_2 = 6\pi \left( \frac{4}{9} \right), S_3 = 6\pi \left( \frac{4}{9} \right)^2$$

$$\sum_{k=1}^{\infty} S_k = \left( \frac{6\pi}{1 - \frac{4}{9}} \right) = \frac{54\pi}{5}$$

$$\Rightarrow \frac{5}{\pi} \sum_{k=1}^{\infty} S_k = \frac{5}{\pi} \cdot \frac{54\pi}{5} = 54$$

## PHYSICS

### SECTION - A

**Multiple Choice Questions:** This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

**Choose the correct answer :**

26. Three infinitely long wires with linear charge density  $\lambda$  are placed along the  $x$ -axis,  $y$ -axis and  $z$ -axis respectively. Which of the following denotes an equipotential surface?
- (1)  $xy + yz + zx = \text{constant}$
  - (2)  $xyz = \text{constant}$
  - (3)  $(x^2 + y^2)(y^2 + z^2)(z^2 + x^2) = \text{constant}$
  - (4)  $(x + y)(y + z)(z + x) = \text{constant}$

**Answer (3)**

**Sol.** Potential due to an infinite wire is  $V = 2k \ln r$ , where  $r$  is the distance from the wire.

Taking the point in space  $P(x, y, z)$

Distance from wire along  $x$ -axis is  $r_x = \sqrt{y^2 + z^2}$

Distance from wire along  $y$ -axis is  $r_y = \sqrt{x^2 + z^2}$

Distance from wire along  $z$ -axis is  $r_z = \sqrt{x^2 + y^2}$

$\Rightarrow$  Potential at  $P$  due to wire along  $x$ -axis is

$$V_x = 2k \ln r$$

Potential at  $P$  due to wire along  $y$ -axis is

$$V_y = 2k \ln r$$

Potential at  $P$  due to wire along  $z$ -axis is

$$V_z = 2k \ln r$$

$\Rightarrow$  Net potential at  $P = V = V_x + V_y + V_z$

$$\text{or } V = 2k\lambda \ln r_x + 2k\lambda \ln r_y + 2k\lambda \ln r_z$$

$$\text{i.e. } V = 2k \ln(r_x r_y r_z)$$

$$\text{or } V = 2k\lambda \ln(\sqrt{y^2 + z^2} \sqrt{z^2 + x^2} \sqrt{x^2 + y^2})$$

$$= k\lambda \ln(y^2 + z^2)(z^2 + x^2)(x^2 + y^2)$$

$\Rightarrow$  For equipotential surface

$$(x^2 + y^2)(y^2 + z^2)(z^2 + x^2) = \text{constant}$$

27. Choose the correct nuclear process from the below options

[ $p$ : proton,  $n$ : neutron,  $e^-$ : electron,  $e^+$ : positron,  $\nu$ : neutrino,  $\bar{\nu}$ : antineutrino]

- (1)  $n \rightarrow p + e^- + \bar{\nu}$
- (2)  $n \rightarrow p + e^+ + \bar{\nu}$
- (3)  $n \rightarrow p + e^- + \nu$
- (4)  $n \rightarrow p + e^+ + \nu$

**Answer (3)**

**Sol.** For all nuclear processes, charge must be conserved. Also, when a release of an electron ( $e^-$ ) is always accompanied by a release of an antineutrino ( $\bar{\nu}$ )

Hence,  $n \rightarrow p + e^- + \bar{\nu}$  is the correct answer.

28. Consider a long thin conducting wire carrying a uniform current  $I$ . A particle having mass " $M$ " and charge " $q$ " is released at a distance " $a$ " from the wire with a speed  $v_0$  along the direction of current in the wire. The particle gets attracted to the wire due to magnetic force. The particle turns round when it is at distance  $x$  from the wire. The value of  $x$  is

[ $\mu_0$  is vacuum permeability]

$$(1) a \left[ 1 - \frac{mv_0}{qIa\mu_0} \right] \quad (2) \frac{a}{2}$$

$$(3) a \left[ 1 - \frac{mv_0}{2qIa\mu_0} \right] \quad (4) ae \frac{-4\pi mv_0}{qIa\mu_0}$$

**Answer (4)**

**Sol.** Let the velocity of the particle make an angle  $\theta$  with initial direction when it is at a distance  $x$ .



$$\Rightarrow \frac{dx}{dt} = -v_0 \sin \theta \quad \dots (i)$$

$$\text{Also, } \frac{d\theta}{dt} = \frac{q}{m} \left( \frac{\mu_0 I}{2\pi x} \right) \quad \dots (ii)$$

$$\Rightarrow \frac{dx}{d\theta} = \frac{-2\pi m v_0 x \sin \theta}{\mu_0 q I} \quad \dots \text{dividing eq(i) with eq(ii)}$$

$$\text{or } \int_a^x \frac{dx}{x} = \frac{-2\pi m v_0}{\mu_0 q I} \int_0^\theta \sin \theta d\theta$$

$$\ln \frac{x}{a} = \frac{-4\pi m v_0}{\mu_0 q I}$$

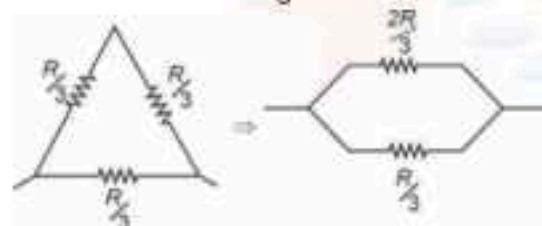
$$\text{Or } x = ae^{\frac{-4\pi m v_0}{\mu_0 q I}}$$

29. A wire of resistance  $R$  is bent into an equilateral triangle and an identical wire is bent into a square. The ratio of resistance between the two end points of an edge of the triangle to that of the square is

- (1) 32/27                      (2) 27/32  
(3) 8/9                        (4) 9/8

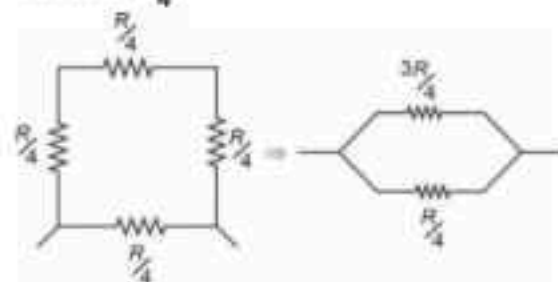
**Answer (1)**

**Sol.** For the wire bent into an equilateral triangle, each side has a resistance  $\frac{R}{3}$ .



$$R_{eq} = \frac{\left( \frac{2R}{3} \parallel \frac{R}{3} \right) \parallel \left( \frac{2R}{3} \parallel \frac{R}{3} \right)}{3} = \frac{2R}{9} \quad \dots \text{(lets say)}$$

For the wire bent into a square, each side has a resistance  $\frac{R}{4}$ .



$$R_{eq} = \frac{\left( \frac{3R}{4} \parallel \frac{R}{4} \right) \parallel \left( \frac{3R}{4} \parallel \frac{R}{4} \right)}{4} = \frac{3R}{16} \quad \dots \text{(lets say)}$$

$$\Rightarrow \frac{R_1}{R_2} = \frac{9}{3R} = \frac{32}{27}$$

30. A proton of mass ' $m_p$ ' has same energy as that of a photon of wavelength ' $\lambda$ '. If the proton is moving at non-relativistic speed, then ratio of its de Broglie wavelength to the wavelength of photon is:

- (1)  $\frac{1}{2c} \sqrt{\frac{E}{m_p}}$                       (2)  $\frac{1}{c} \sqrt{\frac{E}{m_p}}$   
(3)  $\frac{1}{c} \sqrt{\frac{E}{2m_p}}$                         (4)  $\frac{1}{c} \sqrt{\frac{2E}{m_p}}$

**Answer (3)**

**Sol.** Energy of photon =  $E = \frac{hc}{\lambda}$

$$\Rightarrow \text{Wavelength of photon} = \lambda = \frac{hc}{E}$$

$$\text{Energy of proton} = E = \frac{1}{2} m_p v^2 = \frac{p^2}{2m_p}$$

$$\Rightarrow \text{Linear momentum of proton} = P = \sqrt{2m_p E}$$

Or de-Broglie wavelength of proton

$$= \lambda_p = \frac{h}{P} = \frac{h}{\sqrt{2m_p E}}$$

$$\text{Ratio } \frac{\lambda_p}{\lambda} = \frac{h}{\sqrt{2m_p E}} \cdot \frac{E}{hc}$$

$$= \frac{1}{c} \sqrt{\frac{E}{2m_p}}$$

31. A Carnot engine ( $E$ ) is working between two temperatures 473 K and 273 K. In a new system two engines – engine  $E_1$  works between 473 K to 373 K and engine  $E_2$  works between 373 K to 273 K. If  $\eta_{12}$ ,  $\eta_1$  and  $\eta_2$  are the efficiencies of the engines  $E$ ,  $E_1$  and  $E_2$  respectively, then

- (1)  $\eta_{12} = \eta_1 + \eta_2$                       (2)  $\eta_{12} \geq \eta_1 + \eta_2$   
(3)  $\eta_{12} < \eta_1 + \eta_2$                       (4)  $\eta_{12} = \eta_1 \eta_2$

**Answer (3)**

**Sol.** Efficiencies of a carnot engine  $\eta = 1 - \frac{T_{\text{sink}}}{T_{\text{source}}}$

$$\Rightarrow \eta_1 = 1 - \frac{373 \text{ K}}{473 \text{ K}} = \frac{100}{473}$$

$$\eta_2 = 1 - \frac{273 \text{ K}}{373 \text{ K}} = \frac{100}{373}$$

$$\eta_{12} = 1 - \frac{273 \text{ K}}{473 \text{ K}} = \frac{100}{473}$$

$$\eta_{12} - \eta_1 = \frac{100}{473} - \frac{100}{473} = \frac{100}{473} < \frac{100}{373}$$

$$\Rightarrow \eta_{12} - \eta_1 < \eta_2$$

$$\text{or } \eta_{12} < \eta_1 + \eta_2$$

32. Due to presence of an em-wave whose electric component is given by  $E = 100 \sin(\omega t - kx) \text{ NC}^{-1}$ , a cylinder of length 200 cm holds certain amount of em-energy inside it. If another cylinder of same length but half diameter than previous one holds same amount of em-energy, the magnitude of the electric field of the corresponding em-wave should be modified as

- (1)  $50 \sin(\omega t - kx) \text{ NC}^{-1}$     (2)  $400 \sin(\omega t - kx) \text{ NC}^{-1}$   
(3)  $25 \sin(\omega t - kx) \text{ NC}^{-1}$     (4)  $200 \sin(\omega t - kx) \text{ NC}^{-1}$

**Answer (4)**

**Sol.** Energy density of an  $EM_{\text{wave}} = \frac{1}{2} E_0^2$ , where  $E_0$  is the amplitude of the wave.

Since total energy is same for both cylinders

$$\left(\frac{1}{2} \epsilon E_1^2\right) \pi R_1^2 L_1 = \left(-\epsilon E_2^2\right) \pi R_2^2 L_2$$

$$\Rightarrow E_1^2 R_1^2 L_1 = E_2^2 R_2^2 L_2$$

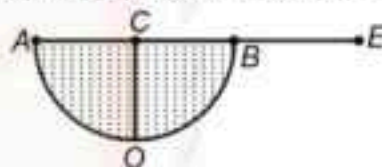
$$\text{or } E_2 = \frac{E_1 R_1}{R_2} \sqrt{\frac{L_1}{L_2}} = \frac{100 \text{ d}}{(d/2)} \sqrt{\frac{L_1}{L_2}} = 200 \text{ N/C}$$

$$[L_1 L_2 = 200 \text{ cm}]$$

$\Rightarrow$  The amplitude of corresponding EM wave is 200 N/C

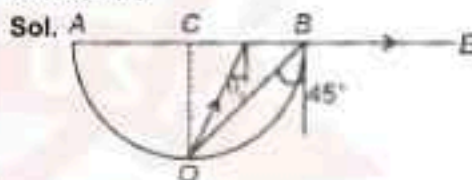
or the wave is  $E = 200 \sin(\omega t - kx) \text{ NC}^{-1}$

33. A hemispherical vessel is completely filled with a liquid of refractive index  $\mu$ . A small coin is kept at the lowest point ( $O$ ) of the vessel as shown in figure. The minimum value of the refractive index of the liquid so that a person can see the coin from point  $E$  (at the level of the vessel) is \_\_\_\_\_.



- (1)  $\sqrt{2}$                                       (2)  $\sqrt{3}$   
(3)  $\frac{3}{2}$                                       (4)  $\frac{\sqrt{3}}{2}$

**Answer (1)**



For the rays from coin to reach the point  $E$ , the refracted rays must grazing the surface, i.e. they must be incident at critical angle  $\theta_c$  inside the liquid.

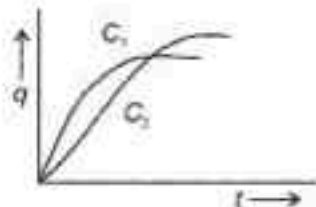
$$\mu = \frac{1}{\sin \theta_c}$$

$\mu$  is minimum when  $\theta_c$  is maximum.

Maximum value of  $\theta_c = 45^\circ$

$\Rightarrow \mu$  has a minimum value of  $\sqrt{2}$ .

34. Two capacitors  $C_1$  and  $C_2$  are connected in parallel to a battery. Charge-time graph is shown below for the two capacitors. The energy stored with them are  $U_1$  and  $U_2$ , respectively. Which of the given statements is true?



- (1)  $C_2 > C_1, U_2 < U_1$       (2)  $C_1 > C_2, U_1 > U_2$   
(3)  $C_1 > C_2, U_1 < U_2$       (4)  $C_2 > C_1, U_2 > U_1$

**Answer (4)**

**Sol.** For a capacitor at steady state

$$q = CV \text{ and } U = \frac{1}{2} CV^2$$

Since  $C_1$  and  $C_2$  are connected in parallel,  $V_1 = V_2$ .

Also from graph  $q_1 < q_2$

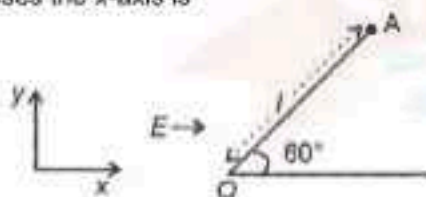
$$\Rightarrow C_1 V_1 < C_2 V_2$$

i.e.  $C_1 < C_2$  or  $C_2 > C_1$

$$\frac{U_1}{U_2} = \frac{C_1 V_1^2}{C_2 V_2^2} = \frac{C_1}{C_2} < 1$$

or  $U_1 < U_2$  or  $U_2 > U_1$

35. A particle of mass ' $m$ ' and charge ' $q$ ' is fastened to one end ' $A$ ' of a massless string having equilibrium length  $l$ , whose other end is fixed at point ' $O$ '. The whole system is placed on a frictionless horizontal plane and is initially at rest. If uniform electric field is switched on along the direction as shown in figure, then the speed of the particle when it crosses the  $x$ -axis is



- (1)  $\sqrt{\frac{qEl}{m}}$                       (2)  $\sqrt{\frac{qEl}{4m}}$   
(3)  $\sqrt{\frac{qEl}{2m}}$                       (4)  $\sqrt{\frac{2qEl}{2m}}$

**Answer (1)**

**Sol.** From work energy theorem,

Work done by electric force = change in kinetic energy

$$\text{or } qE \left( l - \frac{l}{2} \cos 60^\circ \right) = \frac{1}{2} mv^2$$

$$\Rightarrow \frac{qEl}{2} = \frac{1}{2} mv^2$$

$$\text{or } v = \sqrt{\frac{qEl}{m}}$$

36. In the experiment for measurement of viscosity ' $\eta$ ' of given liquid with a ball having radius  $R$ , consider following statements.

- A. Graph between terminal velocity  $V$  and  $R$  will be a parabola.  
B. The terminal velocities of different diameter balls are constant for a given liquid.  
C. Measurement of terminal velocity is dependent on the temperature.  
D. This experiment can be utilized to assess the density of a given liquid.  
E. If balls are dropped with some initial speed, the value of  $\eta$  will change.

Choose the correct answer from the options given below:

- (1) B, D and E only      (2) A, C and D only  
(3) A, B and E only      (4) C, D and E only

**Answer (2)**

**Sol.** The terminal velocity of ball of radius  $R$  inside a liquid of viscosity  $\eta$  can be written as

$$V_T = \frac{2R^2 g}{9\eta} (\sigma - \rho), \text{ where } \sigma \text{ is the density of ball and } \rho \text{ is the density of the liquid.}$$

Hence,

A is correct since  $V_T \propto R^2$  gives a parabola on a graph

C is correct since  $V_T \propto \frac{1}{\eta}$  and  $\eta$  varies with temperature

D is correct since  $V_T \propto (\sigma - \rho)$  i.e., varies with density of liquid.

37. Consider following statements:
- Surface tension arises due to extra energy of the molecules at the interior as compared to the molecules at the surface, of a liquid.
  - As the temperature of liquid rises, the coefficient of viscosity increases.
  - As the temperature of gas increases, the coefficient of viscosity increases.
  - The onset of turbulence is determined by Reynold's number.
  - In a steady flow two stream lines never intersect.

Choose the correct answer from the options given below:

- (1) A, D, E only                      (2) C, D, E only  
(3) A, B, C only                      (4) B, C, D only

**Answer (2)**

**Sol.** Surface tension arises due to extra energy of the molecules at the surface as compared at the interior of a liquid, The coefficient of viscosity for a liquid decreases with rise in temperature where as it increases for gases with increase in temperature. The flow is turbulent for a Reynold's number greater than 2000. Stream lines never intersect in a steady flow.

38. Which of the following circuits has the same output as that of the given circuit?



- 
- 
- 
- 

**Answer (3)**

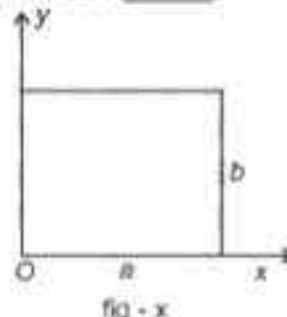
**Sol.**  $Y = \overline{A \cdot B} \cdot \overline{A \cdot B}$

$$= \overline{A(\overline{B} + B)}$$

$$= \overline{A} \text{ as } (\overline{B} + B) = 1$$

i.e. NOT gate with A as input

39. The centre of mass of a thin rectangular plate (fig - x) with sides of length a and b, whose mass per unit area ( $\sigma$ ) varies as  $\sigma = \frac{\sigma_0 x}{ab}$  (where  $\sigma_0$  is a constant), would be \_\_\_\_\_.



- (1)  $\left(\frac{a}{2}, \frac{b}{2}\right)$                       (2)  $\left(\frac{1}{3}a, \frac{2}{3}b\right)$   
(3)  $\left(\frac{2}{3}a, \frac{2}{3}b\right)$                       (4)  $\left(\frac{2}{3}a, \frac{1}{2}b\right)$

**Answer (4)**

**Sol.**  $dm = \sigma dA$

$$= \sigma(dx)(dy) = \frac{\sigma_0 x}{ab}(dx)(dy)$$

$$x_{com} = \frac{\int x dm}{\int dm} = \frac{\int x \left(\frac{\sigma_0 x}{ab}\right)(dx)(dy)}{\int \left(\frac{\sigma_0 x}{ab}\right)(dx)(dy)}$$

$$= \frac{\int_0^a x^2 dx \int_0^b dy}{\int_0^a x dx \int_0^b dy} = \frac{2a}{3}$$

$$y_{com} = \frac{\int y dm}{\int dm} = \frac{\int y \left(\frac{\sigma_0 x}{ab}\right)(dx)(dy)}{\int \left(\frac{\sigma_0 x}{ab}\right)(dx)(dy)}$$

$$= \frac{\int_0^a x dx \int_0^b y dy}{\int_0^a x dx \int_0^b dy} = \frac{b}{2}$$

i.e.,  $r_{com} = \left(\frac{2a}{3}, \frac{b}{2}\right)$

40. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**  
**Assertion A:** A sound wave has higher speed in solids than gases.

**Reason R:** Gases have higher value of Bulk modulus than solids.

In the light of the above statements, choose the **correct** answer from the options given below

- (1) **A** is true but **R** is false  
(2) **A** is false but **R** is true  
(3) Both **A** and **R** are true and **R** is the correct explanation of **A**  
(4) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**

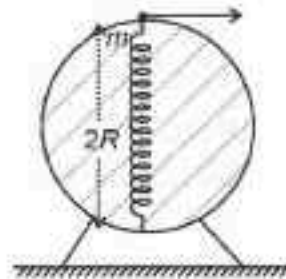
**Answer (1)**

**Sol.** Speed of sound in a medium depends on inertial and elastic properties as  $v = \sqrt{\frac{B}{\rho}}$  for gases and

$v = \sqrt{\frac{Y}{\rho}}$  for solids. Since the elastic property of solid happens to be many folds greater than that of gases, the speed of sound in solids is higher than in gases.

Also, bulk modulus of gases varies between 0 and  $\infty$  ( $B \rightarrow \nu$ )  $\frac{dP}{dV}$  hence reason is false.

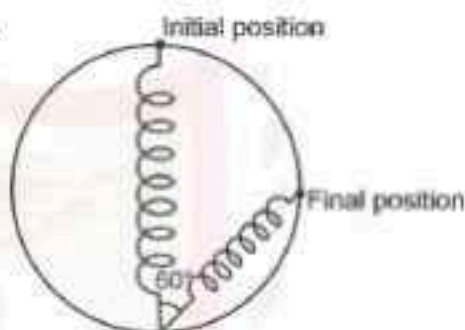
41. A bead of mass ' $m$ ' slides without friction on the wall of a vertical circular hoop of radius ' $R$ ' as shown in figure. The bead moves under the combined action of gravity and a massless spring ( $k$ ) attached to the bottom of the hoop. The equilibrium length of the spring is ' $R$ '. If the bead is released from top of the hoop with (negligible) zero initial speed, velocity of bead, when the length of spring becomes ' $R$ ', would be (spring constant is ' $K$ ',  $g$  is acceleration due to gravity)



- (1)  $\sqrt{2Rg + \frac{4kR^2}{m}}$       (2)  $\sqrt{2Rg + \frac{kR^2}{m}}$   
(3)  $\sqrt{3Rg + \frac{kR^2}{m}}$       (4)  $2\sqrt{gR + \frac{kR^2}{m}}$

**Answer (3)**

**Sol.**



$$\begin{aligned} \text{Work done by gravity} &= mg(2R - R\cos 60^\circ) \\ &= \frac{3mgR}{2} \end{aligned}$$

$$\begin{aligned} \text{Work done by spring} &= -\frac{1}{2}k(0R^2 - R^2) \\ &= \frac{1}{2}kR^2 \end{aligned}$$

Net work = change in kinetic energy

$$\text{i.e. } \frac{3mgR}{2} + \frac{kR^2}{2} = \frac{1}{2}mv^2$$

$$\text{or } v^2 = 3gR + \frac{kR^2}{m}$$

$$\text{or } v = \sqrt{3gR + \frac{kR^2}{m}}$$

42. Given below are two statements : one is labelled as **Assertion A** and the other is labelled as **Reason R**

**Assertion A** : In a central force field, the work done is independent of the path chosen.

**Reason R** : Every force encountered in mechanics does not have an associated potential energy.

In the light of the above statements, choose the **most appropriate** answer from the options given below

- (1) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**
- (2) **A** is true but **R** is false
- (3) Both **A** and **R** are true but **R** is the correct explanation of **A**
- (4) **A** is false but **R** is true

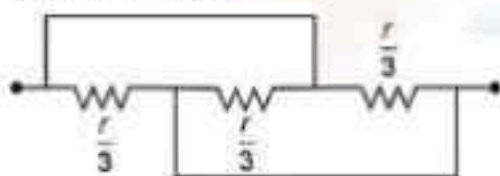
**Answer (1)**

**Sol.** Assertion is correct as central forces are conservative in nature, *i.e.* work done is independent of path.

Reason is true as some forces in mechanics like, friction are non-conservative because work done depends on path and only conservative forces have an associated potential energy.

Also, reason does not explain assertion.

43. Find the equivalent resistance between two ends of the following circuit

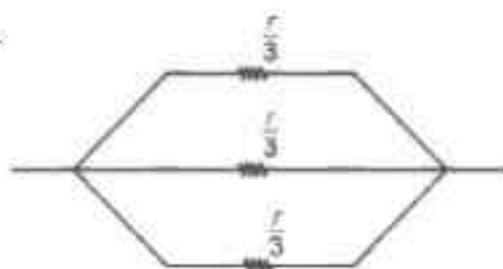


- (1)  $\frac{r}{6}$
- (2)  $\frac{r}{9}$
- (3)  $r$
- (4)  $\frac{r}{3}$

**Answer (2)**

**Sol.** The three resistors are in parallel

*i.e.*



$$\Rightarrow \frac{1}{R_{eq}} = \frac{1}{\frac{r}{3}} + \frac{1}{\frac{r}{3}} + \frac{1}{\frac{r}{3}}$$

$$\text{Or } R_{eq} = \frac{r}{9}$$

44. A thin prism  $P_1$  with angle  $4^\circ$  made of glass having refractive index 1.54, is combined with another thin prism  $P_2$  made of glass having refractive index 1.72 to get dispersion without deviation. The angle of the prism  $P_2$  is degrees is

- (1)  $\frac{16}{3}$
- (2) 1.5
- (3) 4
- (4) 3

**Answer (4)**

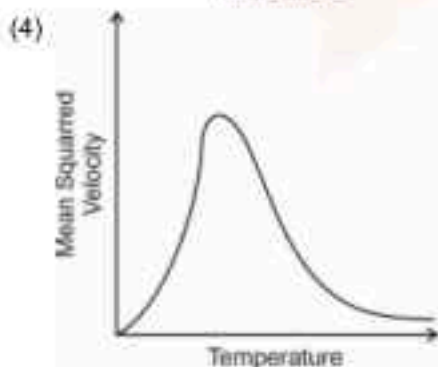
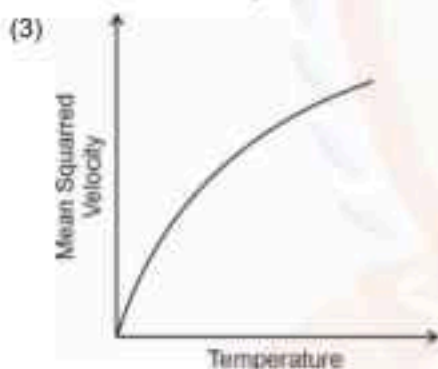
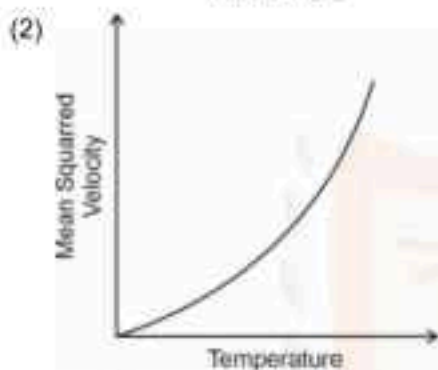
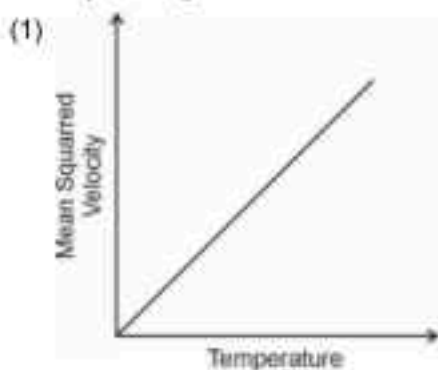
**Sol.** For dispersion without deviation,

$$(\mu_1 - 1)A_1 = (\mu_2 - 1)A_2$$

$$\Rightarrow (1.54 - 1)4^\circ = (1.72 - 1)A_2$$

$$\text{Or } A_2 = \frac{0.54}{0.72} \times 4 = 3^\circ$$

45. For a particular ideal gas which of the following graphs represents the variation of mean squared velocity of the gas molecules with temperature?



**Answer (1)**

**Sol.** Mean squared velocity =  $(v_{rms})^2 = \frac{3RT}{M}$

i.e. (mean squared velocity)  $\propto$  (Temperature)

Hence, graph is a straight line.

**SECTION - B**

**Numerical Value Type Questions:** This section contains 5 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.

46. A double slit interference experiment performed with a light of wavelength 600 nm forms an interference fringe pattern on a screen with 10<sup>th</sup> bright fringe having its centre at a distance of 10 mm from the central maximum. Distance of the centre of the same 10<sup>th</sup> bright fringe from the central maximum when the source of light is replaced by another source of wavelength 660 nm would be \_\_\_\_\_ mm.

**Answer (11)**

**Sol.** Position of the  $n^{\text{th}}$  bright fringe w.r.t. central maxima

in a YDSE is  $y_n = \frac{\lambda D}{d}$

$$\Rightarrow \frac{y'_{10}}{y_{10}} = \frac{\lambda'}{\lambda}$$

or  $y'_{10} = y_{10} \frac{\lambda'}{\lambda}$

$$= \left( \frac{660 \text{ nm}}{600 \text{ nm}} \right) 10 \text{ mm}$$

$$= 11 \text{ mm}$$

47. In a measurement, it is asked to find modulus of elasticity per unit torque applied on the system. The measured quantity has dimension of  $[M^a L^b T^c]$ . If  $b = 3$ , then value of  $c$  is \_\_\_\_\_.

**Answer (0)**

**Sol.** [Modulus of elasticity] =  $ML^{-1}T^{-2}$

$$[\text{Torque}] = ML^2T^{-2}$$

[Modulus of elasticity per unit torque]

$$= \frac{ML^{-1}T^{-2}}{ML^2T^{-2}} L^{-3}$$

48. Two iron solid discs of negligible thickness have radii  $R_1$  and  $R_2$  and moment of inertia  $I_1$  and  $I_2$ , respectively. For  $R_2 = 2R_1$ , then ratio of  $I_1$  and  $I_2$  would be  $1/x$ , where  $x =$  \_\_\_\_\_.

**Answer (16)**

**Sol.** Given  $\frac{M_1}{M_2} = \frac{\pi R_1^2}{\pi R_2^2} = \left(\frac{R_1}{R_2}\right)^2 = \frac{1}{4}$

$$\frac{I_1}{I_2} = \frac{M_1}{M_2} \times \left(\frac{R_1}{R_2}\right)^2 = \frac{1}{4} \times \frac{1}{4} = \frac{1}{16} = \frac{1}{x}$$

$$x = 16$$

49. A tiny metallic rectangular sheet has length and breadth of 5 mm and 2.5 mm, respectively. Using a specially designed screw gauge which has pitch of 0.75 mm and 15 divisions in the circular scale, you are asked to find the area of the sheet. In this measurement, the maximum fractional error will be

$$\frac{x}{100} \text{ where } x \text{ is } \underline{\hspace{2cm}}$$

**Answer (3)**

**Sol.** Least count of screw gauge =

$$\frac{\text{Pitch}}{\text{Number of circular scale divisions}}$$

$$\Rightarrow LC = \frac{0.75 \text{ mm}}{15} = 0.05 \text{ mm}$$

Length,  $l = 5 \text{ mm}$

Breadth,  $b = 2.5 \text{ mm}$

Area =  $lb$

$$\Rightarrow \% \text{ error in area} = (\% \text{ error in length}) + (\% \text{ error in breadth}) = \frac{\Delta l}{l} \times 100 + \frac{\Delta b}{b} \times 100$$

$$= \frac{0.05}{5} \times 100 + \frac{0.05}{2.5} \times 100$$

$$= 3\%$$

50. The moment of inertia of a solid disc rotating along its diameter is 2.5 times higher than the moment of inertia of a ring rotating in similar way. The moment of inertia of a solid sphere which has same radius as the disc and rotating in similar way, is  $n$  times higher than the moment of inertia of the given ring. Here,  $n =$  \_\_\_\_\_.

Consider all the bodies have equal masses.

**Answer (4)**

**Sol.** Given,

$$M_{\text{disc}} = M_{\text{ring}} = M_{\text{sphere}} = m \text{ (let's say) and } R_{\text{sphere}} = R_{\text{disc}}$$

$$I_{\text{disc}} = 2.5 I_{\text{ring}}$$

$$\frac{mR_{\text{disc}}^2}{4} = 2.5 \frac{mR_{\text{ring}}^2}{2}$$

$$\Rightarrow R_{\text{disc}}^2 = 5 R_{\text{ring}}^2$$

$$I_{\text{sphere}} = n I_{\text{ring}}$$

$$\frac{2}{5} mR_{\text{disc}}^2 = n \frac{mR_{\text{ring}}^2}{2}$$

$$n = 4$$

## CHEMISTRY

### SECTION - A

**Multiple Choice Questions:** This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

**Choose the correct answer :**

51. Match the LIST-I with LIST-II

	LIST-I (Redox Reaction)		LIST-II (Type of Redox Reaction)
A.	$\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \xrightarrow{\Delta} \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{l})$	I.	Disproportionation reaction
B.	$2\text{NaH}(\text{s}) \xrightarrow{\Delta} 2\text{Na}(\text{s}) + \text{H}_2(\text{g})$	II.	Combination reaction
C.	$\text{V}_2\text{O}_5(\text{s}) + 5\text{Ca}(\text{s}) \xrightarrow{\Delta} 2\text{V}(\text{s}) + 5\text{CaO}(\text{s})$	III.	Decomposition reaction
D.	$2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\Delta} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$	IV.	Displacement reaction

Choose the **correct** answer from the options given below:

- (1) A-II, B-III, C-I, D-IV    (2) A-IV, B-I, C-II, D-III  
 (3) A-III, B-IV, C-I, D-II    (4) A-II, B-III, C-IV, D-I

**Answer (4)**

- Sol.** (A)  $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \xrightarrow{\Delta} 2\text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{l}) \Rightarrow$  combination reaction  
 (B)  $2\text{NaH}(\text{s}) \xrightarrow{\Delta} 2\text{Na}(\text{s}) + \text{H}_2(\text{g}) \Rightarrow$  Decomposition reaction  
 (C)  $\text{V}_2\text{O}_5(\text{s}) + 5\text{Ca}(\text{s}) \xrightarrow{\Delta} 2\text{V}(\text{s}) + 5\text{CaO}(\text{s}) \Rightarrow$  Displacement reaction  
 (D)  $2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\Delta} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g}) \Rightarrow$  Disproportionation reaction

52. Given below are two statements:

**Statement I:** D-glucose pentaacetate reacts with 2, 4-dinitrophenylhydrazine

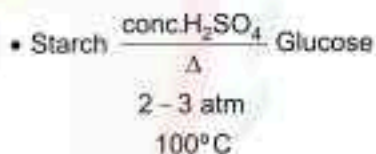
**Statement II:** Starch, on heating with concentrated sulfuric acid at 100°C and 2-3 atmosphere pressure produces glucose.

In the light of the above statements, choose the **correct** answer from the options given below

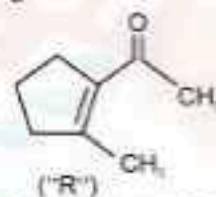
- (1) Both Statement I and Statement II are true  
 (2) Statement I is true but Statement II is false  
 (3) Statement I is false but Statement II is true  
 (4) Both Statement I and Statement II are false

**Answer (3)**

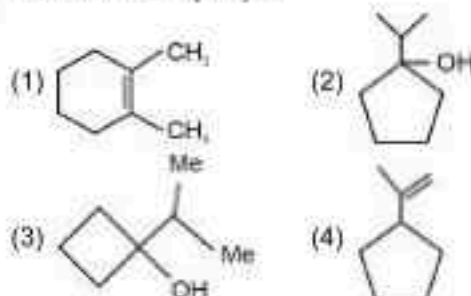
**Sol.** • D-glucose pentaacetate do not react with 2,4 DNP



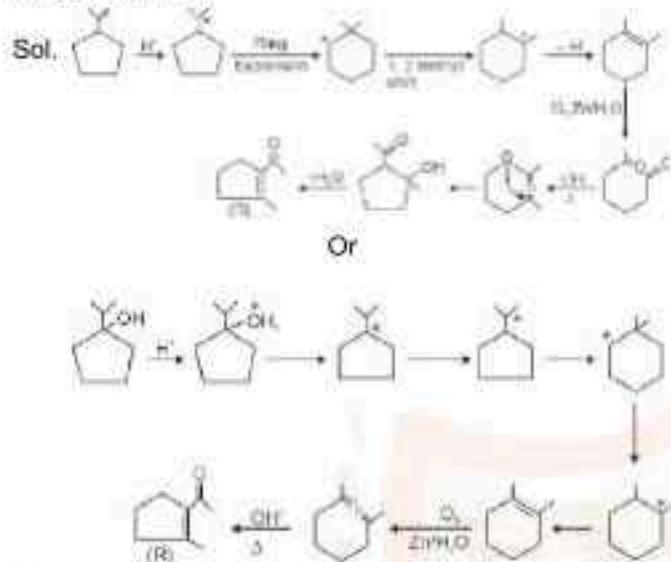
53. A molecule ("P") on treatment with acid undergoes rearrangement and gives ("Q"). ("Q") on ozonolysis followed by reflux under alkaline condition gives ("R"). The structure of ("R") is given below.



The structure of ("P") is



Answer (2 and 4)



54.

$[A]_0 / \text{mol L}^{-1}$	$t_{1/2} / \text{min}$
0.100	200
0.025	100

For a given reaction  $R \rightarrow P$ ,  $t_{1/2}$  is related to  $[A]_0$  as

given in table.

Given;  $\log 2 = 0.30$

Which of the following is true?

- A. The order of the reaction is  $\frac{1}{2}$ .
- B. If  $[A]_0$  is 1M, then  $t_{1/2}$  is  $200\sqrt{10}$  min
- C. The order of the reaction changes to 1 if the concentration of reactant changes from 0.100 M to 0.500 M.
- D.  $t_{1/2}$  is 800 min for  $[A]_0 = 1.6$  M

Choose the **correct** answer from the options given below:

- (1) A, B and D only      (2) A and C only  
(3) C and D only      (4) A and B only

Answer (1)

Sol.  $t_{1/2} \propto (C_0)^{1-n}$

$$\frac{t_1}{t_2} = \left( \frac{C_1}{C_2} \right)^{1-n}$$

$$\Rightarrow \frac{200}{100} = \left( \frac{0.100}{0.025} \right)^{1-n}$$

$$\Rightarrow 2 = (4)^{1-n}$$

$$(1-n) = \frac{1}{2}$$

$$n = \frac{1}{2}$$

For  $n = \frac{1}{2}$

$$\frac{-dA}{dt} = k(A)^{\frac{1}{2}}$$

$$\int_{C_0}^C \frac{dA}{(A)^{\frac{1}{2}}} = - \int_0^t k dt$$

$$\Rightarrow 2A^{\frac{1}{2}} = -kt$$

$$\Rightarrow \sqrt{C} - \sqrt{C_0} = \frac{-kt}{2}$$

$$\sqrt{C} - \sqrt{C_0} = \frac{kt}{2}$$

For  $C_0 = 0.1 \Rightarrow t_{1/2} = 200$  min.

$$\sqrt{\frac{C_0}{2}} = \sqrt{C_0} - \frac{kt}{2}$$

$$\frac{kt}{2} = \sqrt{C_0} - \sqrt{\frac{C_0}{2}}$$

$$\frac{kt}{2} = \sqrt{C_0} \left( \frac{\sqrt{2}-1}{\sqrt{2}} \right)$$

$$t_{1/2} = \frac{2\sqrt{C_0}}{k} \left( \frac{\sqrt{2}-1}{\sqrt{2}} \right)$$

$$200 = \frac{2\sqrt{0.1}}{k} \left( \frac{\sqrt{2}-1}{\sqrt{2}} \right)$$

$$k = \frac{\sqrt{0.1}(\sqrt{2}-1)}{100 \left( \frac{\sqrt{2}-1}{\sqrt{2}} \right)}$$

For  $C_0 = 1M$

$$t_{\frac{1}{2}} = \frac{2 \times 100(\sqrt{2})}{\sqrt{0.1}(\sqrt{2}-1)} \times \frac{(\sqrt{2}-1)}{\sqrt{2}}$$

$$\Rightarrow 200\sqrt{10} \text{ min.}$$

$\Rightarrow$  B is correct

C is incorrect

For  $C_0 = 1.6 M$

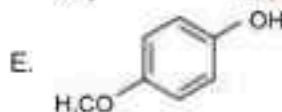
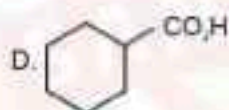
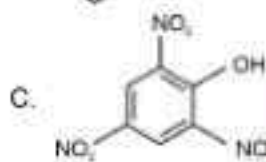
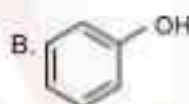
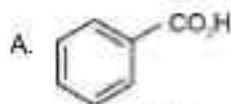
$$t_{\frac{1}{2}} = \frac{2\sqrt{1.6}(\sqrt{2})(\sqrt{2}-1) \times 100}{\sqrt{0.1}(\sqrt{2}-1)(\sqrt{2})}$$

$$t_{\frac{1}{2}} = 400 \times 2 \text{ min}$$

$$t_{\frac{1}{2}} = 800 \text{ min}$$

A, B and D are correct

55. The compounds that produce  $CO_2$  with aqueous  $NaHCO_3$  solution are:



Choose the **correct** answer from the options given below:

- (1) A, C and D only      (2) A and C only  
(3) A and B only      (4) A, B and E only

**Answer (1)**

**Sol.** A, C, D produces  $CO_2$  with  $NaHCO_3$  as these are stronger acid as compared to  $H_2CO_3$  (carbonic acid)

56. The incorrect decreasing order of atomic radii is  
(1)  $Si > P > Cl > F$       (2)  $Mg > Al > C > O$   
(3)  $Be > Mg > Al > Si$       (4)  $Al > B > N > F$

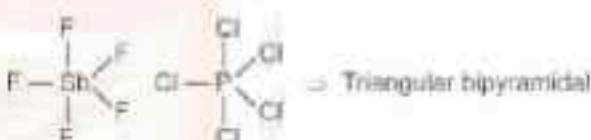
**Answer (3)**

**Sol.** Correct order of atomic radii  $Be < Mg > Al > Si$ .

57. ~~The~~ molecules having square pyramidal geometry

- (1)  $BrF_5$  &  $PCl_5$       (2)  $BrF_5$  &  $XeOF_4$   
(3)  $SbF_5$  &  $XeOF_4$       (4)  $SbF_5$  &  $PCl_5$

**Answer (2)**



58. ~~pressure of the system is increased to 2 atm~~ If keeping temperature constant, then identify correct observation from following

- (1) Volume of system increases  
(2) The solid phase (ice) disappears completely  
(3) Liquid phase disappears completely  
(4) The amount of ice decreases

**Answer (2)**

**Sol.**



On increasing pressure to 2 atm, freezing point of ice will decrease and solid phase (ice) will disappear completely.

59. In a multielectron atom, which of the following orbitals described by three quantum numbers will have same energy in absence of electric and magnetic fields?

- A.  $n = 1, l = 0, m_l = 0$   
B.  $n = 2, l = 0, m_l = 0$   
C.  $n = 2, l = 1, m_l = 1$   
D.  $n = 3, l = 2, m_l = 1$   
E.  $n = 3, l = 2, m_l = 0$

Choose the **correct** answer from the options given below:

- (1) A and B only                      (2) C and D only  
(3) B and C only                      (4) D and E only

**Answer (4)**

**Sol.**  $(n + l)$  must be same.

	$(n + l)$
A	1
B	2
C	3
D	5
E	5

$(n + l)$  value for D and E is same.

60. Which out of the following oxidation reactions are  $\text{Cr}_2\text{O}_7^{2-}$  and  $\text{KMnO}_4$  in acidic medium?

- A.  $\text{I}^- \rightarrow \text{I}_2$                               B.  $\text{S}^{2-} \rightarrow \text{S}$   
C.  $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$                         D.  $\text{I}^- \rightarrow \text{IO}_3^-$   
E.  $\text{S}_2\text{O}_3^{2-} \rightarrow \text{SO}_4^{2-}$

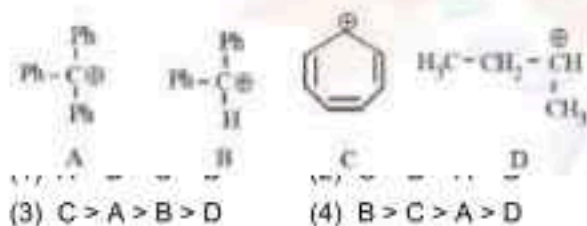
Choose the **correct** answer from the option given below.

- (1) A, B and C only                      (2) A, D and E only  
(3) C, D and E only                      (4) B, C and D only

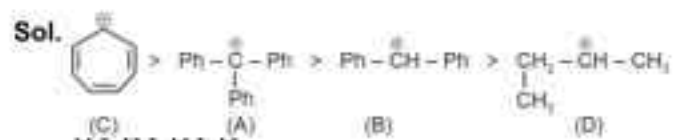
**Answer (1)**

**Sol.**  $\text{I}^-$  gets converted in  $\text{I}_2$  by acidified  $\text{KMnO}_4$ .

61. The correct order of stability of following



**Answer (3)**



62. Consider 'n' is the number of lone pair of electrons present in the equatorial position of the most stable structure of  $\text{ClF}_3$

The ions from the following with 'n' number of unpaired electrons are

- A.  $\text{V}^{3+}$                                       B.  $\text{Ti}^{3+}$   
C.  $\text{Cu}^{2+}$                                       D.  $\text{Ni}^{2+}$   
E.  $\text{Ti}^{2+}$

Choose the **correct** answer from the options given below:

- (1) A, D and E Only                      (2) B and C Only  
(3) A and C Only                        (4) B and D Only

**Answer (1)**

**Sol.**  $n = 2$

$\text{V}^{3+} : 4s^0 3d^2$  ( $n = 2$ )  $\Rightarrow$  A is correct

$\text{Ti}^{3+} : 4s^0 3d^1$  ( $n = 1$ )  $\Rightarrow$  B is not correct

$\text{Cu}^{2+} : 4s^0 3d^9$  ( $n = 1$ )  $\Rightarrow$  C is not correct

$\text{Ni}^{2+} : 4s^0 3d^8$  ( $n = 2$ )  $\Rightarrow$  D is correct

$\text{Ti}^{2+} : 4s^0 3d^2$  ( $n = 2$ )  $\Rightarrow$  E is correct

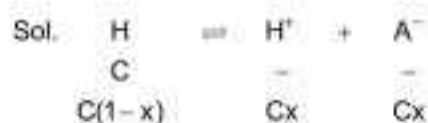
A, D and E are correct

Option (1) is correct

63. A weak acid HA has degree of dissociation x. Which option gives the correct expression of  $(\text{pH} - \text{pK}_a)$ ?

- (1)  $\log(1 + 2x)$   
(2)  $\log\left(\frac{1-x}{x}\right)$   
(3)  $\log\left(\frac{x}{1-x}\right)$   
(4) 0

**Answer (3)**



$$K_a = \frac{[\text{H}^+](x)}{\text{C}(1-x)}$$

$$[\text{H}^+] = K_a \frac{(1-x)}{x}$$

$$\log \text{H}^+ = \log K_a + \log \left( \frac{1-x}{x} \right)$$

$$-\log \text{H}^+ = -\log K_a - \log \frac{1-x}{x}$$

$$\text{pH} = \text{p}K_a - \log \frac{1-x}{x}$$

$$\text{pH} - \text{p}K_a = \log \frac{x}{1-x}$$

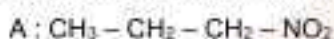
64. ~~Suppose~~ **Suppose** A and B in the following reactions,



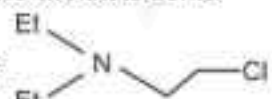
- (1)  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{NO}_2$ ,  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CN}$
- (2)  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{ONO}$ ,  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CN}$
- (3)  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{ONO}$ ,  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{NC}$
- (4)  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{NO}_2$ ,  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{NC}$

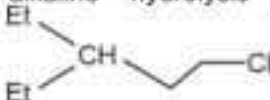
**Answer (4)**

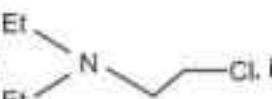
Sol.  $\text{NO}_2^-$  and  $\text{CN}^-$  are ambidentate nucleophiles



65. Given below are two statements:

**Statement I:**  will undergo alkaline hydrolysis at a faster rate than



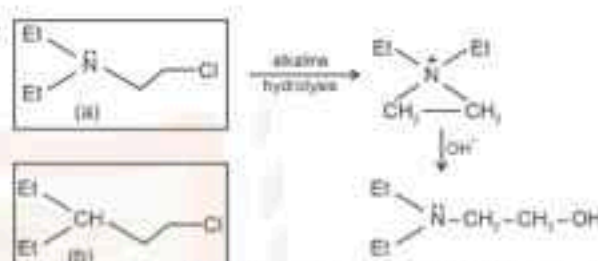
**Statement II:** , intramolecular substitution takes place first by involving lone pair

In the light of the above statements, choose the most appropriate answer from the options given below

- (1) Statement I is incorrect but Statement II is correct
- (2) Statement I is correct but Statement II is incorrect
- (3) Both Statement I and Statement II are incorrect
- (4) Both Statement I and Statement II are correct

**Answer (4)**

Sol.



intra molecular substitution.

66. What is the freezing point depression constant of a solvent, 50 g of which contain 1 g non volatile solute (molar mass 256 g mol<sup>-1</sup>) and the decrease in freezing point is 0.40 K?
- (1) 1.86 K kg mol<sup>-1</sup>
  - (2) 5.12 K kg mol<sup>-1</sup>
  - (3) 4.43 K kg mol<sup>-1</sup>
  - (4) 3.72 L kg mol<sup>-1</sup>

**Answer (2)**

Sol.  $\Delta T_f = 0.40 \text{ K}$

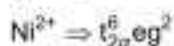
$$(0.40) = (K_f) \left( \frac{1000}{(256)(50)} \right)$$

$$\boxed{(K_f) = 5.12 \text{ K kg mole}^{-1}}$$

67. The metal ion whose electronic configuration is not affected by the nature of the ligand and which gives a violet colour in non luminous flame under hot condition in borax bead test is
- (1)  $\text{Ni}^{2+}$
  - (2)  $\text{Mn}^{2+}$
  - (3)  $\text{Cr}^{3+}$
  - (4)  $\text{Ti}^{3+}$

**Answer (1)**

Sol.  $Ni^{2+}$  gives violet colour with borax bead test in non-luminous flame under hot conditions.  $Ni^{2+}$  has  $d^8$  configuration which does not depend on nature of ligand present in octahedral field.



68. Both acetaldehyde and acetone (individually) undergo which of the following reactions?

- A. Iodoform Reaction
- B. Cannizzaro Reaction
- C. Aldol Condensation
- D. Tollen's Test
- E. Clemmensen Reduction

Choose the **correct** answer from the options given below:

- (1) B, C and D only      (2) A, B and D only
- (3) A, C and E only      (4) C and E only

**Answer (3)**

**Sol.**

	Name of Reaction	Acetaldehyde	Acetone
A.	Iodoform reaction	✓	✓
B.	Cannizzaro reaction	×	×
C.	Aldol reaction	✓	✓
D.	Tollen's test	✓	×
E.	Clemmensen reduction	✓	✓

69. Given below are two statements :

**Statement I** : In the oxalic acid vs  $KMnO_4$  (in the presence of dil  $H_2SO_4$ ) titration the solution needs to be heated initially to  $60^\circ C$ , but no heating is required in Ferrous ammonium sulphate (FAS) vs  $KMnO_4$  titration (in the presence of dil  $H_2SO_4$ )

**Statement II** : In oxalic acid vs  $KMnO_4$  titration, the initial formation of  $MnSO_4$  takes place at high

temperature, which then acts as catalyst for further reaction. In the case of FAS vs  $KMnO_4$ , heating oxidizes  $Fe^{2+}$  into  $Fe^{3+}$  by oxygen of air and error may be introduced in the experiment.

In the light of the above statements, choose the **correct** answer from the options given below

- (1) Both Statement I and Statement II are false
- (2) Statement I is true but Statement II is false
- (3) Statement I is false but Statement II is true
- (4) Both Statement I and Statement II are true

**Answer (4)**

**Sol.** Heating is required in oxalic acid filtration due to High activation energy.

Heating is not required in FAS vs  $KMnO_4$  titration because  $Fe^{2+}$  will get converted into  $Fe^{3+}$  by oxygen of air and error may be introduced in the experiment.

Both Statement-I and Statement-II are correct.

70. Consider the following elements In, Ti, Al, Pb, Sn and Ge.

The most stable oxidation states of elements with highest and lowest first ionisation enthalpies, respectively, are

- (1) +2 and +3      (2) +4 and +3
- (3) +4 and +1      (4) +1 and +4

**Answer (2)**

**Sol.** In, Ti, Al, Pb, Sn, Ge

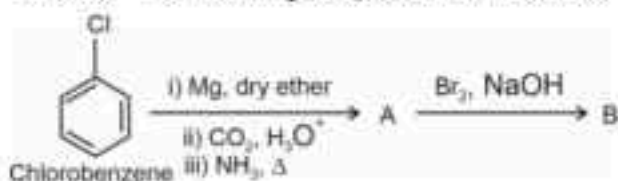
Lowest first I.E. = In  $\Rightarrow$  most stable oxidation state  $\Rightarrow$  +3

Highest first I.E. = Ge  $\Rightarrow$  most stable oxidation state  $\Rightarrow$  +4

### SECTION - B

**Numerical Value Type Questions:** This section contains 5 Numerical based questions. The answer to each question should be rounded off to the nearest integer.

71. Consider the following sequence of reactions:



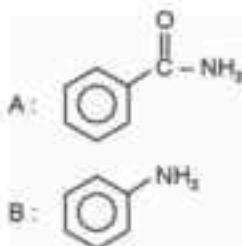
11.25 mg of chlorobenzene will produce  $\text{---} \times 10^{-1}$  mg of product B. conversion.)

(Consider the reactions result in complete 16, 14 and 35.5 g mol<sup>-1</sup> respectively)

[Given molar mass of C, H, O, N and Cl as 12, 1,

**Answer (93)**

**Sol.**



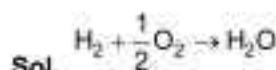
$$\text{moles of chlorobenzene} = \frac{11.25 \times 10^{-3}}{112.5} = 10^{-4}$$

$$\text{moles of product B} = 10^{-4}$$

$$\text{mass of product B} = (10^{-4}) \times 93 = 93 \times 10^{-1} \text{ milligram}$$

72. The formation enthalpies,  $\Delta H_f$  for  $\text{H}_{2(g)}$  and  $\text{O}_{(g)}$  are 220.0 and 250.0 kJ mol<sup>-1</sup>, respectively, at 298.15 K, and  $\Delta H_f$  for  $\text{H}_2\text{O}_{(g)}$  is -242.0 kJ mol<sup>-1</sup> at the same temperature. The average bond enthalpy of the O-H bond in water at 298.15 K is \_\_\_\_\_ kJ mol<sup>-1</sup> (nearest integer).

**Answer (466)**



$$(-242) = 440 + 250 - 2x$$

$$2x = 440 + 250 + 242$$

$$x = 466$$

73. Quantitative analysis of an organic compound (X) shows following % composition.

$$\text{C} : 14.5\% \quad \text{Cl} : 64.46\%$$

$$\text{H} : 1.8\%$$

(Empirical formula mass of the compound (X) is  $\text{---} \times 10^{-1}$ )

(Given molar mass in g mol<sup>-1</sup> of C : 12, H : 1, O : 16, Cl : 35.5)

**Answer (1655)**

**Sol.**  $\text{C} \left( \frac{14.5}{12} \right) \text{H} \left( \frac{1.8}{1} \right) \text{Cl} \left( \frac{64.46}{35.5} \right) \text{O} \left( \frac{19.24}{16} \right)$

$$\text{C}_{1.21} \text{H}_{1.8} \text{Cl}_{1.81} \text{O}_{1.21}$$

$$\Rightarrow \text{C}_1 \text{H}_{1.5} \text{Cl}_{1.5} \text{O}_1$$

$$\Rightarrow \text{C}_2 \text{H}_3 \text{Cl}_3 \text{O}_2$$

$$\begin{aligned} \text{Empirical formula mass} &= 165.5 \frac{\text{gm}}{\text{mole}} \\ &= 1655 \times 10^{-1} \end{aligned}$$

74. The molarity of a 70% (mass/mass) aqueous solution of a monobasic acid (X) is  $\text{---} \times 10^{-1}$

M(Nearest integer)

[Given : Density of aqueous solution of (X) is 1.25 g mL<sup>-1</sup>

Molar mass of the acid is 70 g mol<sup>-1</sup>]

**Answer (125)**

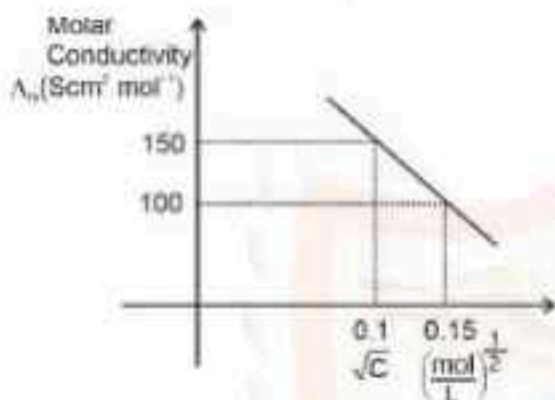
**Sol.** Moles of solute =  $\frac{70}{70} = 1$

$$\text{Volume of solution} = \frac{100}{1.25} = 80 \text{ mL}$$

$$M = \frac{1}{80} \times 1000 = 12.5$$

$$M = 125 \times 10^{-1}$$

75. Given below is the plot of the molar conductivity vs  $\sqrt{\text{concentration}}$  for KCl in aqueous solution.



If, for the higher concentration of KCl solution, the resistance of the conductivity cell is 100 W, then the resistance of the same cell with the dilute solution is 'x' W

The value of x is \_\_\_\_\_ (Nearest integer)

**Answer (150)**

**Sol.**  $100 = \rho \left( \frac{l}{A} \right)$

For  $\sqrt{C} = 0.15$   
 $l_m = 100$

$$100 = \frac{k \times 1000}{0.15 \times 0.15}$$

$$k = \frac{15 \times 15}{10^5} = \frac{225}{10^5}$$

$$\rho = \frac{10^5}{225}$$

$$100 = \frac{10^5}{225} \left( \frac{l}{A} \right)$$

$$\frac{l}{A} = \frac{225}{1000}$$

For  $\sqrt{C} = 0.1$

$$l_m = 150$$

$$150 = \frac{k \times 1000}{10^{-2}}$$

$$k = 15 \times 10^{-4}$$

$$\rho = \frac{10^4}{15}$$

$$R = \rho \left( \frac{l}{A} \right)$$

$$R = \frac{10^4}{15} \times \frac{225}{1000}$$

$$R \Rightarrow 150 \text{ W}$$

